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A solution of 0.20 mole of NaBr and 0.20 mole of MgBr₂ in 2.0 dm³ of water is to be analysed. How many moles of Pb(NO₃)₂ must be added to precipitate all the bromide as insoluble PbBr₂. A. 0.30 mol. B. 0.10 mol. ... WAEC 2020 Chemistry Theory Questions PAPER 2 (ESSAY) SECTION A. 1. Identify the solid remaining when each of the following ...

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(2) Determination of concentration of liquid solutes (stock solution) given the density, w/v, w/w), specific gravity, relative molecular mass, molar mass, and % purity. (3) Primary standard, secondary standard and standardized solution. WAEC Syllabus For Chemistry | 5.0 ENERGY AND ENERGY CHANGES (a) Energy changes in physical and isolated systems.

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WAEC Chemistry Questions and Answers for 2020/2021. 100.0g of KClO₃ was added to 40.0 cm³ of water to give a saturated solution at 298K. If the solubility of the salt is 20.0 moldm⁻³ at 298K, what percentage of the salt is left undissolved?

WAEC Chemistry Questions and Answers for 2020/2021 (Theory ...

Chemistry in waec is divided into objective (50 questions) and theory (this is divided into section A and section B). Your success in waec Chemistry begin if you can make sure to score minimum of 35 out of the 50 questions in the objective.

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