

## Monatomic Ions Answers Key

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### Answers To Monatomic Ions - Teacher Worksheets

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### Weebly

Monatomic Ions Use a periodic table to complete the table below: Ion Formula Element Name sodium bromine magnesium chlorine oxygen boron lithium neon phosphorus Element Symbol Ion Name 2. 3. 4. 5. 6. 8. 9. 10. aluminum 1 1 -calcium 12. iodine 1 3, nitrogen 14. cesium 15. sulfur 16. fluorine 17. potassium 18. barium 19. hydrogen 20. helium

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Based on your knowledge of monatomic ions, propose an explanation for the net charge on a polyatomic ion. Your answer should include a discussion of subatomic particles. Somewhere in the ion there are either extra (in the case of an ion) or fewer electrons (in the case of a cation) compared to the total number of protons in the atoms involved. 7.

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Section B Write the formula of the ionic compounds containing polyatomic ions 1. lead (II) chlorate  $Pb(ClO_3)_2$  2. strontium acetate  $Sr(C_2H_3O_2)_2$  3. zinc phosphate  $Zn_3(PO_4)_2$  4. ammonium cyanide  $NH_4CN$  5. sodium carbonate  $Na_2CO_3$  6. lead (IV) dichromate  $Pb(Cr_2O_7)_2$  7. copper (I) sulfite  $Cu_2SO_3$  8. cadmium phosphate  $Cd_3(PO_4)_2$  9. tin ( II ) bicarbonate ...

### Answer Key Worksheet 5.1 Naming and Writing Ionic ...

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Write out the full electron configuration for each of the following atoms and for the monatomic ion found in binary ionic compounds containing the element: Al; Br; Sr; Li; As; S; Answer a. Al:  $1s^2 2s^2 2p^6 3s^2 3p^1$ . Al  $3+$ :  $1s^2 2s^2 2p^6$ . Answer b. Br:  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^5$ . Br $-$ :  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6$ . Answer c. Sr:  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 5s^2$ . Sr  $2+$ :  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6$

**3.1: Electron Configurations (Problems) - Chemistry LibreTexts**

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**Ionic Bonding Worksheet 1**

6. Use the table of ions in Model 1 to answer the following questions: a. In the compound zinc phosphide, what is the charge on the zinc ion? b. In the compound zinc phosphide, what is the charge on the phosphide ion?  
7. Explain why a 3 to 2 ratio of ions is necessary for the compound zinc phosphide. POGILTM Activities for High School Chemistry

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Numerical Solutions The answer is (b) Law of conservation of mass. The number of grams of reactants (32.0 g of methane and 128.0 g of oxygen = 160.0 g total) is equal to the number of grams of product (88.0 g of carbon dioxide and 72.0 g of water = 160.0 g total). The answer is 18.00 g of water.

**2.E: Atoms, Molecules, and Ions (Exercises) - Chemistry ...**

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